

DP-003-2016006

Seat No.

B. Sc. (Sem. VI) (CBCS) (W.E.F. 2019) Examination

March - 2022

Chemistry: C-601

(Inorganic Chemistry & Industrial Chemistry)

(New Course)

Faculty Code: 003

Subject Code: 2016006

Time : 2.30 Hours] [Total Marks : 70

Instruction: (1) All questions are compulsory.

- (2) All questions carry 14 marks each.
- 1 (A) Answer the following.

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- (1) What is zero point energy for one dimensional box ?
- (2) Define addition of operators.
- (3) Give equation to calculate energy for a particle in cubic box.
- (4) Write down Schrodinger's equation for a particle moving in one dimensional box.
- (B) Answer any one.

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- (1) Normalize the wave function $\Psi = Ne^{im\phi} (0 \le \phi \le 2\pi)$.
- (2) Calculate degeneracy of energy level $E = \frac{9h^2}{8ma^2}$ for a particle moving in a cubic box of length 'a'.
- (C) Answer any one.

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- (1) Explain 'Commutator of Operators'
- (2) Give transformation of cartesian coordinates into polar coordinates.

	(D)	Ans	wer any one.	5
		(1)	Write Schrodinger's equation in polar coordinates and derive $R_{(r)}$, $\theta_{(\theta)}$ and $\phi_{(\phi)}$ equation by variable separation method.	
		(2)	Discuss linear polyenses as one dimansional be system and explain calculation of its c-c bond length	
2	(A)	Answer the following. 4		
		(1)	What is the relation between χ_{para} and Temperatu	re.
		(2)	Give equation of magnetic induction.	
		(3)	Define, 'Unit Pole'	
		(4)	What is the value of permeability for para and diamagnetic compound?	nd
	(B)	Ans	wer any one.	2
		(1)	Explain, 'Diamagnetism'	
		(2)	Explain the term, 'Magnetic Permeability'	
	(C)	Ans	wer any one.	3
		(1)	Explain about magnetic susceptibility and its types	•
		(2)	Explain, "Ferro and Anti ferro magnetic compunds"	•
	(D)	Answer any one. 5		5
		(1)	Explain types of magnetism and effect of temperating magnetic bodies.	ture on
		(2)	Derive equation for the diamagnetic momentum and li the characteristics of diamagnetic susceptibility.	st
3	(A)	Answer the following. 4		
		(1)	Give one example of each mononuclear and poly-nuclear metal carbonyl.	
		(2)	Define π -acid ligand	
		(3)	What is stand oil ?	
		(4)	List the uses of animal fat.	
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	(B)	Answer any one.					
		(1)	List any five properties of fat.				
		(2)	Give chemical reaction for the hydrogenation of oil in presence of Raney Nickel.				
	(C)	Ans	wer any.	3			
		(1)	Explain Dry Process of Hydrogenation of oil.				
		(2)	Define saponification value and give equation to calculate it.				
(D)	(D)	Answer any one.					
		(1)	Discuss the structure of $Co_2(CO)_8$				
		(2)	Describe the importance of IR spectra in the study of metal carbonyls.				
4 (A	(A)	Ans	Answer the following.				
		(1)	List any five sources of water pollution.				
		(2)	Define COD.				
		(3)	What is Pollutant ?				
		(4)	What are the sources of production of SO_x and NO_x	?			
(B)	Answer any one.						
		(1)	Give chain reaction of CFC depleting Ozone at stratosphere.				
		(2)	List the Green House Gases along with their % contribution towards GWP.				
	(C)	Ans	wer any one.	3			
		(1)	Write a short note on "Petrochemical Smog".				
		(2)	What is thermal pollution? Discuss the effects of thermal pollution.				
	(D)	Ans	wer any one.	5			
		(1)	Discuss "Green House Effect".				
		(2)	Discuss how acid rain occurs with proper chemical reactions involved.				
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5 (A) Answer the following. 4 (1) What is Soap? (2) What is semipolar detergent? Give types of surfactants besed on lyophobic part present in it. (4) Define sulphur soap and neem soap. (B) Answer any one. 2 (1) Give basic principle of soap manufacture with its chemical reaction. (2) List the row material used for manufacture of soap. (C) Answer any one. 3 (1) Explain cationic and anionic detergents. Discuss recovery of Glycerine from spent lye. 5 (D) Answer any one. Discuss synthesis of anionic detergent by Alfol method. **(1)**

Describe Hot process of soap manufacturing.

(2)